

Structure of the Product from the Reaction Between Methyl 1-Methyl-1,2,3,4,9,10-hexahydrophenanthren-1-yl Ketone and Ethyl Orthoformate

BY PHILIP J. COX

School of Pharmacy, Robert Gordon's Institute of Technology, Schoolhill, Aberdeen AB9 1FR, Scotland

AND CHRISTOPHER J. GILMORE AND GEORGE A. SIM

Chemistry Department, University of Glasgow, Glasgow G12 8QQ, Scotland

(Received 22 December 1986; accepted 24 March 1987)

Abstract. (4 β ,7 α ,10 α S*,13S*)-5,6,7,7a,9,10,11,12-Octahydro-7a,13-dimethyl-8H-4b,9-(epoxymethano)-cyclopenta[*j*]phenanthren-8-one, $C_{20}H_{24}O_2$, $M_r = 296.43$, monoclinic, $P2_1$, $a = 7.285(1)$, $b = 14.052(2)$, $c = 7.699(2)$ Å, $\beta = 94.54(2)^\circ$, $V = 786(1)$ Å³, $Z = 2$, $D_x = 1.25$ Mg m⁻³, Cu $K\alpha$, $\lambda = 1.5418$ Å, $\mu = 0.54$ mm⁻¹, $F(000) = 320$, $T = 293$ K, $R = 0.036$ for 1577 reflections with $I > 2.5\sigma(I)$. The X-ray analysis confirms the structure proposed on spectroscopic grounds for the compound obtained from the reaction between methyl 1-methyl-1,2,3,4,9,10-hexahydrophenanthren-1-yl ketone and ethyl orthoformate, involving attack on C(4a) and C(10a) of the former. The aromatic C–C bond lengths are 1.382(4)–1.399(4) Å, mean 1.391 Å. The C(sp³)–C(sp³) bond lengths range from 1.521(3) to 1.559(2) Å, mean 1.534 Å, the longest bonds being associated with the fully substituted C atoms [C(1), C(4a), C(10a)].

Table 1. Fractional atomic coordinates ($\times 10^4$) and equivalent isotropic thermal parameters (Å² $\times 10^3$)

$$U_{eq} = (U_{11}U_{22}U_{33})^{1/3}$$

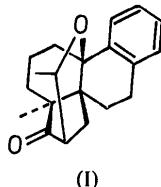
	x	y	z	U_{eq}
O(1)	6703(2)	8954	12898(2)	60(1)
O(2)	8880(2)	7512(2)	9990(2)	37(1)
C(1)	5556(2)	8713(2)	9856(3)	40(1)
C(2)	6620(3)	9532(2)	9109(3)	48(1)
C(3)	8513(3)	9284(2)	8523(3)	48(1)
C(4)	8449(3)	9374(2)	7435(3)	41(1)
C(4a)	7636(2)	7547(2)	8427(2)	33(1)
C(5)	9424(3)	6134(2)	7475(3)	44(1)
C(5a)	7746(2)	6619(2)	7418(2)	36(1)
C(6)	9626(3)	5309(2)	6528(3)	55(1)
C(7)	8126(4)	4954(2)	5507(3)	59(1)
C(8)	6451(3)	5418(2)	5449(3)	53(1)
C(8a)	6236(3)	6254(2)	6400(2)	42(1)
C(9)	4370(3)	6700(3)	6374(3)	52(1)
C(10)	4367(3)	7701(2)	7167(3)	47(1)
C(10a)	5623(2)	7753(2)	8851(2)	36(1)
C(11)	6298(3)	8415(2)	11704(3)	42(1)
C(12)	6365(3)	7330(2)	11810(3)	44(1)
C(13)	5104(3)	7032(2)	10227(3)	43(1)
C(14)	8309(3)	7009(2)	11461(2)	43(1)
C(15)	9772(4)	7209(3)	12936(3)	62(1)
C(16)	3566(3)	9062(2)	10062(4)	58(1)

Experimental. Colourless crystal of title compound (I) from pentane, dimensions 0.2 × 0.2 × 0.4 mm. Enraf–Nonius diffractometer, graphite monochromator, Cu $K\alpha$ radiation, generator settings 43 kV, 26 mA. Cell dimensions from setting angles of 25 independent reflections with θ 14–20°; 1760 intensities surveyed in the range θ 2–72°; h −8–8, k 0–17, l 0–9; $\omega/2\theta$ scan, scan width (0.80 + 0.14 tan θ)°; scan rate 1.20–6.67° min⁻¹; max. scan time 120 s; 1577 independent reflections with $I > 2.5\sigma(I)$. Two reference reflections

Table 2. Bond lengths (Å) and angles (°) with e.s.d.'s in parentheses

O(1)–C(11)	1.209(3)	C(5)–C(6)	1.384(3)
O(2)–C(4a)	1.449(2)	C(5a)–C(8a)	1.397(3)
O(2)–C(14)	1.425(2)	C(6)–C(7)	1.387(4)
C(1)–C(2)	1.526(3)	C(7)–C(8)	1.382(4)
C(1)–C(10a)	1.559(2)	C(8)–C(8a)	1.399(3)
C(1)–C(11)	1.540(3)	C(8a)–C(9)	1.496(3)
C(1)–C(16)	1.551(3)	C(9)–C(10)	1.532(3)
C(2)–C(3)	1.525(3)	C(10)–C(10a)	1.528(2)
C(3)–C(4)	1.526(3)	C(10a)–C(13)	1.534(2)
C(4)–C(4a)	1.535(2)	C(11)–C(12)	1.527(3)
C(4a)–C(5a)	1.523(2)	C(12)–C(13)	1.526(3)
C(4a)–C(10a)	1.554(2)	C(12)–C(14)	1.530(3)
C(5)–C(5a)	1.397(3)	C(14)–C(15)	1.521(3)
C(1)–C(11)–O(1)	125.4(2)	C(12)–C(11)–O(1)	125.5(2)
C(14)–O(2)–C(4a)	118.3(1)	C(4)–C(4a)–O(2)	101.4(1)
C(5a)–C(4a)–O(2)	109.9(1)	C(10a)–C(4a)–O(2)	111.7(1)
C(12)–C(14)–O(2)	109.0(1)	C(15)–C(14)–O(2)	105.7(2)
C(10a)–C(1)–C(2)	115.4(2)	C(11)–C(1)–C(2)	113.7(2)
C(16)–C(1)–C(2)	107.9(2)	C(3)–C(2)–C(1)	115.9(2)
C(11)–C(1)–C(10a)	101.6(1)	C(16)–C(1)–C(10a)	113.0(2)
C(4a)–C(10a)–C(1)	109.4(1)	C(10)–C(10a)–C(1)	115.0(2)
C(13)–C(10a)–C(1)	102.1(1)	C(16)–C(1)–C(11)	104.7(2)
C(12)–C(11)–C(1)	109.1(2)	C(4)–C(3)–C(2)	111.4(2)
C(4a)–C(4)–C(3)	111.0(2)	C(5a)–C(4a)–C(4)	110.9(1)
C(10a)–C(4a)–C(4)	111.8(1)	C(10a)–C(4a)–C(5a)	110.8(1)
C(5)–C(5a)–C(4a)	118.9(2)	C(8a)–C(5a)–C(4a)	122.0(2)
C(10)–C(10a)–C(4a)	108.9(1)	C(13)–C(10a)–C(4a)	108.1(1)
C(6)–C(5)–C(5a)	121.4(2)	C(8a)–C(5a)–C(5)	119.1(2)
C(7)–C(6)–C(5)	119.2(2)	C(8)–C(8a)–C(5a)	119.1(2)
C(9)–C(8a)–C(5a)	121.9(2)	C(8)–C(7)–C(6)	120.2(2)
C(8a)–C(8)–C(7)	121.0(2)	C(9)–C(8a)–C(8)	119.0(2)
C(10)–C(9)–C(8a)	114.1(2)	C(10a)–C(10)–C(9)	111.2(2)
C(13)–C(10a)–C(10)	113.0(2)	C(12)–C(13)–C(10a)	101.8(1)
C(13)–C(12)–C(11)	102.6(2)	C(14)–C(12)–C(11)	108.1(2)
C(14)–C(12)–C(13)	106.6(2)	C(15)–C(14)–C(12)	114.8(2)

(122 and 212) monitored periodically showed no significant variation in intensity. No absorption correction. Lorentz-polarization correction. Structure determined by direct phasing using *MITHRIL* (Gilmore, 1984). H atoms located in a difference Fourier synthesis. Full-matrix least-squares calculations on *F* with *SHELX76* (Sheldrick, 1976); anisotropic thermal parameters for C and O atoms and isotropic for H. Convergence at $R = 0.036$, $wR = 0.038$, $\Delta/\sigma < 0.1$, with $w = 1/\sigma^2(|F_0|)$. Final $\Delta\rho$ max. 0.15, min. $-0.18 \text{ e } \text{\AA}^{-3}$. Scattering factors for C, H and O atoms were those incorporated in *SHELX76*.



Atomic coordinates are listed in Table 1 and molecular dimensions in Table 2.* Fig. 1, drawn with *ORTEP* (Johnson, 1965), illustrates the molecular structure.

* Lists of structure factors, torsional angles, anisotropic thermal parameters of the C and O atoms, positional and thermal parameters of the H atoms, and bond lengths involving H atoms have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 43908 (17 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

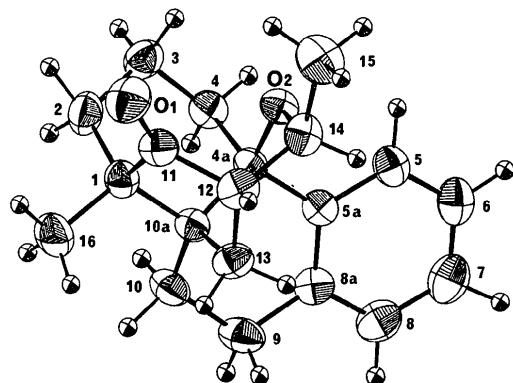


Fig. 1. Molecular structure and atomic numbering (Ghatak *et al.*, 1980). The thermal ellipsoids of the C and O atoms are drawn at the 50% probability level and the H atoms are represented by spheres of radius 0.1 Å.

Related literature. For the preparation of the compound see Ghatak, Sanyal, Ghosh, Sarkar, Raju & Wenkert (1980).

References

- GHATAK, U. R., SANYAL, B., GHOSH, S., SARKAR, M., RAJU, M. S. & WENKERT, E. (1980). *J. Org. Chem.* **45**, 1081–1085.
- GILMORE, C. J. (1984). *J. Appl. Cryst.* **17**, 42–46.
- JOHNSON, C. K. (1965). *ORTEP*. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee, USA.
- SHEDDRICK, G. M. (1976). *SHELX76*. Program for crystal structure determination. Univ. of Cambridge, England.

Acta Cryst. (1987). **C43**, 1643–1645

Structure of 3,3,5,5,7,7-Hexaphenyl-1,2,4,6,3,5,7-tetraazatriphosphhepine Hydrochloride Monohydrate

BY A. W. CORDES

Department of Chemistry and Biochemistry, University of Arkansas, Fayetteville, AR 72701, USA

AND R. T. OAKLEY

Guelph-Waterloo Centre for Graduate Work in Chemistry, Guelph Campus, Department of Chemistry and Biochemistry, University of Guelph, Guelph, Ontario, Canada N1G 2W1

(Received 12 February 1987; accepted 7 April 1987)

Abstract. $\text{C}_{36}\text{H}_{32}\text{N}_4\text{P}_3^+\text{Cl}^- \cdot \text{H}_2\text{O}$, $\text{NHNHPPh}_2\text{NPPh}_2^- \text{NPPh}_2^+$, $M_r = 667.1$, monoclinic, $P2_1/n$, $a = 8.646 (4)$, $b = 20.969 (4)$, $c = 19.276 (3) \text{ \AA}$, $\beta = 96.14 (3)^\circ$, $V = 3475 (3) \text{ \AA}^3$, $Z = 4$, $D_x = 1.27 \text{ g cm}^{-3}$,

$\lambda(\text{Mo } K\alpha) = 0.71073 \text{ \AA}$, $\mu = 2.77 \text{ cm}^{-1}$, $F(000) = 1392$, $T = 293 \text{ K}$, $R = 0.047$ for 3805 unique observed reflections. The highly puckered P_3N_4 ring has $\text{P}-\text{N}$ distances of 1.568 (2) to 1.659 (2) \AA , $\text{N}-\text{N}$ of